## Practice - 18.3 Coulomb's Law

1. What is the repulsive force between two pith balls that are 8.00 cm apart and have equal charges of -30.0 nC?

 $F = \frac{kq}{q^2} \left( \frac{8.99 \times 10^{10} \text{ m}^2}{6^2} \right) 30.0 \times 10^{10} \text{ c}^2$   $= 11.26 \times 10^{10} \text{ N}$ 

2. Two point charges exert a 5.00 N force on each other. What will the force become if the distance between them is increased by a factor of three?

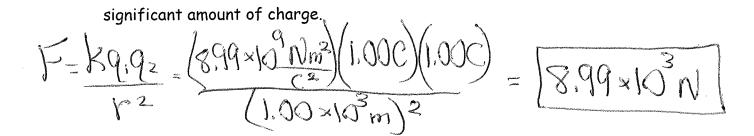
Since  $F = \frac{1}{9}$ , when  $r \uparrow 3x$ ,  $F \downarrow 9x$  $F = \frac{5.00N}{9} = \boxed{0.556N}$ 

3. Two point charges are brought closer together, increasing the force between them by a factor of 25. By what factor was their separation decreased?

When F125x, r15x 5x

4. How far apart must two point charges of 75.0 nC (typical of static electricity) be to have a force of 1.00 N between them?

 $F = kq.q2 = \sqrt{kq.q2} = \sqrt{8.99 \times 10^9 \text{ M/m}^2} / 75.0 \times 10^9 \text{ M/m}^2}$   $= \sqrt{7.11 \times 10^3 \text{ m}} / 7.11 \text{ mm}$ 



5. If two equal charges each of 1 C each are separated in air by a distance of 1 km,

what is the magnitude of the force acting between them? You will see that even at a distance as large as 1 km, the repulsive force is substantial because 1 C is a very

6. A test charge of +2  $\mu$ C is placed halfway between a charge of +6  $\mu$ C and another of +4  $\mu$ C separated by 10 cm. What is the magnitude and direction of the force on the test charge?

$$F_{NET} = F_{6-2} - F_{4-2}$$

$$= \left(8.99 \times 10.00 \text{ m}^{3}\right) \left(6.00 \times 10^{\circ}\right) \left(2.00 \times 10^{\circ}\right) \left(4.00 \times 10^{\circ}\right) \left(4.00 \times 10^{\circ}\right) \left(5.00 \times 10^{\circ}\right)^{2}$$

$$= \left(4.00 \times 10^{\circ}\right) \left(5.00 \times 10^{\circ}\right)^{2}$$

$$= \left(4.00 \times 10^{\circ}\right)^{2}$$

7. Find the ratio of the electrostatic to gravitational force between two electrons.

$$\frac{F_{E}}{F_{G}} = \frac{kq.q^{2}}{6m.m^{2}} = \frac{8.99 \times 10^{9} \times 1.60 \times 10^{19} \times$$

8. A certain five cent coin contains 5.00 g of nickel. What fraction of the nickel atoms' electrons, removed and placed 1.00 m above it, would support the weight of this coin? The atomic mass of nickel is 58.7, and each nickel atom contains 28 electrons and 28 protons.

$$F_{E} = F_{g} \Rightarrow \frac{kq^{2}}{r^{2}} = mg$$

$$9 = \sqrt{\frac{mgr^2}{k}} = \sqrt{\frac{5.00 \times 10^3 kg}{(8.99 \times 10^9 Nm^2)}} = 2.335 \times 100$$

